

N,N'-Bis(3-dimethylaminopropyl)dithiooxamide

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Key indicators

Single-crystal X-ray study
 $T = 294\text{ K}$
 $\text{Mean } \sigma(\text{C-C}) = 0.004\text{ \AA}$
 Disorder in main residue
 R factor = 0.045
 wR factor = 0.151
 Data-to-parameter ratio = 16.7

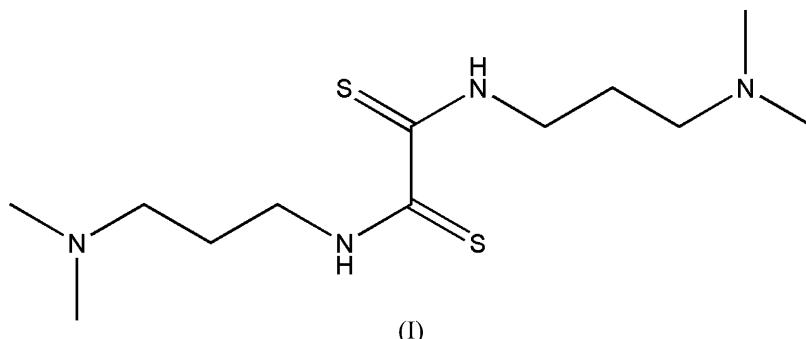
For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

Molecules of the title compound, $C_{12}H_{26}N_4S_2$, have $2/m$ symmetry. Intramolecular N—H···N and N—H···S hydrogen bonds are observed.

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Comment

Substituted dithiooxamides with N-coordinating groups can be used to synthesize multinuclear metal complexes with S- and N-donor atoms. A few studies on such ligands (Hurd *et al.*, 1960; Bermejo *et al.*, 1998; Cui, Zhang, Pan *et al.*, 2005; Cui, Zhang, Song *et al.*, 2005; Cui, Zhang, Zhang *et al.*, 2005) and their dinuclear or trinuclear metal complexes coordinated by substituted dithiooxamides with 2-(aminomethyl)pyridine (Cui *et al.*, 1999), 2-(aminoethyl)pyridine (Cui *et al.*, 2004), amino acids (Veit *et al.*, 1984; Castineiras *et al.*, 2001) and 2-hydroxyethyl (Girerd *et al.*, 1978) have been reported. We report here the X-ray crystal structure determination of the title compound, (I).



The asymmetric unit of (I) contains one quarter-molecule disposed on a mirror plane with a second methyl group generated by reflection, and with the other half generated by twofold rotation symmetry, the axis passing through the inversion centre at the mid-point of the $\text{C}1-\text{C}1^i$ bond [Fig. 1 and Table 1; symmetry code: (i) $-x, -y, -z$]. There are intramolecular N—H···N and N—H···S hydrogen bonds, as detailed in Table 2.

Experimental

Compound (I) was prepared according to the literature procedure of Hurd *et al.* (1961). Ethanol solutions of one molar equivalent of dithiooxamide and two molar equivalents of *N,N*-dimethylaminopropylamine were mixed and stirred at room temperature for 8 h to obtain (I) as an orange powder (yield 34%; m.p. 316–317 K). Single crystals suitable for X-ray analysis were obtained by slow evaporation of a petroleum ether solution of (I). IR (KBr): ν 2943, 2853, 2815, 1519, 1460, 1380, 1245, 1160, 922, 880, 762 cm^{-1} .

Crystal data

$C_{12}H_{26}N_4S_2$
 $M_r = 290.49$
 Tetragonal, $P4_2/mbc$
 $a = 14.4361 (19) \text{ \AA}$
 $c = 8.2304 (15) \text{ \AA}$
 $V = 1715.2 (4) \text{ \AA}^3$
 $Z = 4$
 $D_x = 1.125 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation
 Cell parameters from 2361 reflections
 $\theta = 2.8\text{--}26.2^\circ$
 $\mu = 0.30 \text{ mm}^{-1}$
 $T = 294 (2) \text{ K}$
 Block, orange
 $0.24 \times 0.20 \times 0.14 \text{ mm}$

Data collection

Bruker SMART CCD area-detector diffractometer
 φ and ω scans
 Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)
 $T_{\min} = 0.921$, $T_{\max} = 0.959$
 8983 measured reflections

953 independent reflections
 599 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.046$
 $\theta_{\text{max}} = 26.4^\circ$
 $h = -18 \rightarrow 17$
 $k = -18 \rightarrow 12$
 $l = -8 \rightarrow 10$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.045$
 $wR(F^2) = 0.151$
 $S = 1.04$
 953 reflections
 57 parameters
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0532P)^2 + 1.1916P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.18 \text{ e \AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.24 \text{ e \AA}^{-3}$
 Extinction correction: *SHELXL97*
 Extinction coefficient: 0.067 (6)

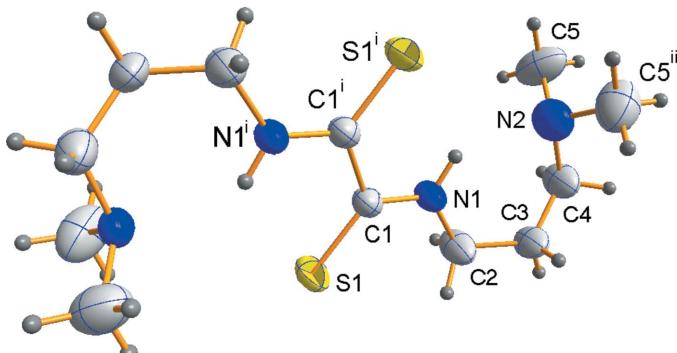


Figure 1

The molecular structure of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 30% probability level. [Symmetry codes: (i) $-x, -y, -z$; (ii) $x, y, 1 - z$.]

Data collection: *SMART* (Bruker, 1997); cell refinement: *SAINT* (Bruker, 1997); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1997); software used to prepare material for publication: *SHELXTL*.

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Table 1
 Selected geometric parameters (\AA , $^\circ$).

S1—C1	1.665 (4)	N2—C4	1.459 (5)
N1—C1	1.306 (5)	N2—C5	1.425 (4)
N1—C2	1.450 (5)		
S1—C1—N1	124.5 (3)	C1—N1—C2	124.3 (3)
S1—C1—C1 ⁱ	121.7 (4)	C4—N2—C5	110.6 (3)
N1—C1—C1 ⁱ	113.8 (4)	C5—N2—C5 ⁱⁱ	111.2 (5)

Symmetry codes: (i) $-x, -y, -z + 1$; (ii) $x, y, -z + 1$.

Table 2
 Hydrogen-bond geometry (\AA , $^\circ$).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
N1—H1A…N2	0.88	2.17	2.851 (4)	134
N1—H1A…S1 ⁱ	0.88	2.41	2.938 (3)	118

Symmetry code: (i) $-x, -y, -z + 1$.

Atom C3 is disordered over the mirror plane and was refined anisotropically with an occupancy factor of 0.50. All H atoms were positioned geometrically and refined using the riding-model approximation, with $\text{N}-\text{H} = 0.88 \text{ \AA}$ and $\text{C}-\text{H} = 0.96 \text{ \AA}$, and with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{N and methylene C})$ or $1.5U_{\text{eq}}(\text{methyl C})$.

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